



Highly bioavailable dust-borne iron delivered to the Southern Ocean during glacial periods

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Changes in bioavailable dust-borne iron (Fe) supply to the iron-limited Southern Ocean may influence climate by modulating phytoplankton growth and CO₂ fixation into organic matter that is exported to the deep ocean. The chemical form (speciation) of Fe impacts its bioavailability, and glacial weathering produces highly labile and bioavailable Fe minerals in modern dust sources. However, the speciation of dust-borne Fe reaching the iron-limited Southern Ocean on glacial–interglacial timescales is unknown, and its impact on the bioavailable iron supply over geologic time has not been quantified. Here we use X-ray absorption spectroscopy on subantarctic South Atlantic and South Pacific marine sediments to reconstruct dust-borne Fe speciation over the last glacial cycle, and determine the impact of glacial activity and glaciogenic dust sources on bioavailable Fe supply. We show that the Fe(II) content, as a percentage of total dust-borne Fe, increases from ~5 to 10% in interglacial periods to ~25 to 45% in glacial periods. Consequently, the highly bioavailable Fe(II) flux increases by a factor of ~15 to 20 in glacial periods compared with the current interglacial, whereas the total Fe flux increases only by a factor of ~3 to 5. The change in Fe speciation is dominated by primary Fe(II) silicates characteristic of glaciogenic dust. Our results suggest that glacial physical weathering increases the proportion of highly bioavailable Fe(II) in dust that reaches the subantarctic Southern Ocean in glacial periods, which represents a positive feedback between glacial activity and cold glacial temperatures.

weathering (10, 11). While glaciogenic sediments and glacial ice core dust have been shown to be highly efficient at fertilizing Fe-limited phytoplankton (10, 17), the impact of glaciogenic versus nonglaciogenic dust-borne Fe speciation on the bioavailable iron supply reaching the Southern Ocean over the last glacial cycle has not been quantified. Instead, all dust-borne Fe is considered equal in biogeochemical models, regardless of its solid-phase speciation (18, 19). In this paper, we use bulk Fe K-edge X-ray absorption spectroscopy (XAS) to determine the speciation—i.e., the average oxidation state [Fe(II)/Fe_{total}] and mineral composition—of dust-borne Fe deposited to Southern Ocean sediment cores over the last glacial cycle. We observe that Fe(II)/Fe_{total} is higher in glacial versus interglacial periods, and that glacially derived primary Fe(II) silicates dominate the Fe deposited to the Southern Ocean during glacial periods. We use microprobe-based X-ray fluorescence (μXRF) and Fe K-edge XAS (μXAS) to confirm the presence of distinct, pure primary Fe(II) particles in the sediments that are physically weathered from bedrock and are not the result of diagenesis. Since previous work has shown that primary Fe(II) silicates are more bioavailable for a given Fe flux than other forms of Fe (10, 11), and the relationship between particulate Fe(II) and bioavailability is linear under Fe limitation (10), we show that primary Fe(II) flux is likely a better estimate of bioavailable Fe supply than dust flux or total Fe flux, and we propose a positive feedback between

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Iron (Fe) fertilization of phytoplankton in the Southern Ocean is thought to contribute to the glacial–interglacial changes in atmospheric CO₂ concentrations (e.g., refs. 1–6). Increased dust-borne Fe deposition and Fe fertilization of phytoplankton in the subantarctic South Atlantic Ocean is associated specifically with the latter half of CO₂ drawdown in the last glacial cycle, as evidenced by positive correlations between productivity proxies, nutrient utilization proxies, and dust/total Fe fluxes measured in marine sediment cores (1, 2, 6). Since John Martin made his Fe hypothesis in 1990 (7) through today, researchers have relied on total dust and total Fe fluxes to marine sediment cores in the Southern Ocean to evaluate the importance of Fe fertilization on geologic timescales (1, 2, 8, 9). The potential additional effect of the chemical form of dust-borne Fe is unknown, despite studies showing that glacial processes impact solid-phase Fe speciation and increase dust-borne Fe bioavailability in modern sources (10–12). Others have observed high dust-borne Fe solubility (13, 14) and high concentrations of leachable, bioavailable Fe²⁺ (15) in Antarctic ice cores at the Last Glacial Maximum (LGM) compared with interglacials, but these studies do not measure solid-phase dust-borne Fe speciation nor evaluate its impact on Fe solubility and bioavailability.

The solid-phase Fe in glaciogenic sediments is more labile and bioavailable than that in nonglaciogenic sediments because it comprises more Fe(II) versus Fe(III) minerals (10–12, 16). These Fe(II) minerals are typically primary Fe(II) silicates that are freshly weathered from bedrock, in contrast with secondary Fe(III) oxyhydroxides that have undergone more chemical

Significance

Dust-borne iron fertilization of Southern Ocean phytoplankton contributes to lower glacial atmospheric CO₂. Previous studies evaluating the impact of dust on climate estimate bioavailable iron using total iron fluxes in sediment cores. Thus, all iron is considered equally bioavailable over geologic time, despite evidence that glaciers mobilize highly bioavailable iron from bedrock, which winds can deliver to the Southern Ocean. Here we reconstruct dust-borne iron speciation over the last glacial cycle, showing that highly bioavailable iron(II) silicate minerals are a greater fraction of total iron reaching the Southern Ocean during glacial periods. The abundance of iron(II) silicates likely controls the bioavailable iron supply to the Southern Ocean and contributes to the previously observed increase in glacial productivity and CO₂ drawdown.

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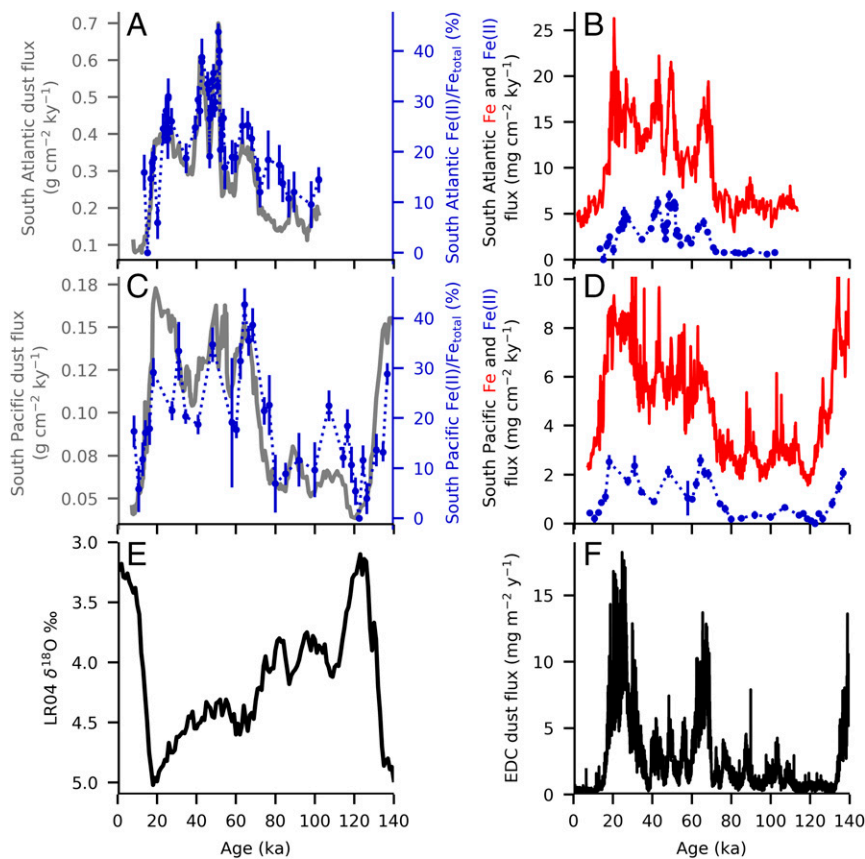


Fig. 2. (A and C) Dust fluxes and $\text{Fe(II)/Fe}_{\text{total}}$ records, with (B and D) total Fe fluxes and Fe(II) fluxes over the last glacial cycle for (A and B) the South Atlantic (TN057-06) and (C and D) the South Pacific (PS75/56-1). (A and C) Dust fluxes are represented with gray solid lines (gray axes). $\text{Fe(II)/Fe}_{\text{total}}$ records are blue circles connected with a dotted line (blue axes), and they are calculated the same way as those in Fig. 1B. The error bars for $\text{Fe(II)/Fe}_{\text{total}}$ are based on the goodness of fit using the best-fit combination of standards, and are produced using Larch (25). (B and D) Fe fluxes are ^{230}Th -normalized (red lines), and LCF-based Fe(II) fluxes (blue circles connected with a dotted line) are the $\text{Fe(II)/Fe}_{\text{total}}$ fraction multiplied by the Fe flux. The error bars for Fe(II) flux were calculated by propagating the errors on $\text{Fe(II)/Fe}_{\text{total}}$, Fe concentration, and ^{230}Th MAR. (E) The LR04 $\delta^{18}\text{O}$ benthic stack (27) (climate proxy) shows the glacial–interglacial cycle for the period of time spanned by the samples in this study. High $\delta^{18}\text{O}$ values indicate cold glacial climates, and vice versa (28). Note that the axis is inverted, by convention. (F) The mineral dust flux to the Antarctic EPICA Dome C (EDC) ice core shows trends in dust supply in this time period (29).

combined with their prevalence in modern glaciogenic dust sources (10), suggest that they are important to phytoplankton in the mixed layer of the ocean. Secondary Fe(II) minerals describe good portions of the low interglacial Fe(II) fluxes (*SI Appendix, Fig. S8*), which is consistent with chemical weathering processes that are more important in interglacial periods (31).

To confirm the presence of distinct primary Fe(II) silicate minerals in the unaltered marine sediment core samples, we use an X-ray microprobe with a $2\ \mu\text{m} \times 2\ \mu\text{m}$ X-ray beam size to identify and characterize individual Fe-rich particles in a glacial (42.7 ka) and interglacial (92.34 ka) sediment sample from the South Atlantic (Fig. 3). The small beam size allows us to probe the speciation of Fe in individual particles in the sediment sample (32), in contrast to bulk XAS, which provides the average Fe speciation and identifies the mix of minerals. Using μXRF maps, we observe large ($\sim 5\ \mu\text{m}$ to $10\ \mu\text{m}$) Fe hotspots over a diffuse Fe signal representing most particles (*SI Appendix, Fig. S9*). We attribute these large hotspots to primary Fe minerals that are ground from bedrock, and we attribute the background to secondary Fe species (clays, Fe oxides) that are predominantly small ($< 2\ \mu\text{m}$) particles and aggregates (33–35). We use μXAS to determine the speciation and mineral composition of the hotspots; LCF at the near-edge region with standard spectra show that parts of these Fe hotspots are $\sim 100\%$ Fe(II) attributed solely to primary Fe(II) silicate minerals (biotite and hornblende were used as representative standards; Fig. 3). The particles in the glacial sediment are more reduced, on the whole, than those in the interglacial sediment (Fig. 3), and the Fe(II) hotspot/ Fe_{total} estimates from the μXRF maps (*SI Appendix, Fig. S9*) are nominally similar to the LCF-based $\text{Fe(II)/Fe}_{\text{total}}$ calculations (Fig. 2). Our identification of biotite- and hornblende-rich particles in the glacial and interglacial sediment provides evidence that distinct Fe(II) primary minerals are deposited to the Southern Ocean

and preserved through the processes of sediment deposition, core collection, and core storage. We show evidence of pure Fe(II) silicates in both glacial and interglacial sediments (Fig. 3), which suggests glaciogenic minerals likely contribute to $\text{Fe(II)/Fe}_{\text{total}}$ throughout the glacial cycle, simply to a greater degree in glacial periods and to a lesser degree in interglacial periods.

We can rule out diagenetic controls on Fe speciation in the cores, because Fe(II) silicates dominate the Fe(II) signal. In marine sediments, the unknown impacts of diagenesis have precluded previous efforts to reconstruct dust-borne Fe speciation and bio-availability in the Southern Ocean over the last glacial cycle, since reducing conditions in the sediment can alter Fe oxidation state and speciation (36). Since primary Fe(II) silicates form only from cooling magma (37) and metamorphic processes above $500\ ^\circ\text{C}$ (38, 39), changes in primary Fe(II) silicate concentrations represent glacial–interglacial changes in dust source [mainly the degree of physical weathering versus chemical weathering (31)] rather than diagenesis. Specifically, the primary Fe(II) silicates that are most commonly identified in sediments, including chlorite, biotite, and hornblende, cannot form under the low temperature and pressure conditions of sediment cores (37–39), and dominate glaciogenic dust sources impacted by physical weathering of bedrock (10, 11, 40). Secondary phyllosilicates (e.g., smectite, kaolinite, glauconite) produced during chemical weathering and diagenesis (41) can also contain some Fe(II) , but they are structurally distinct from primary Fe(II) silicates (shown with biotite and glauconite XAS in Fig. 3) and represent minor contributions to the total Fe(II) signal except in interglacial periods [low dust, low Fe(II) content; *SI Appendix, Fig. S8*]. Iron sulfide minerals (e.g., pyrite, which was used in all LCF analyses), which would likely dominate any authigenic Fe(II) signal (36), contribute only minimally to the mineral composition of all samples as determined using LCF (*SI Appendix, Fig. S8*). However, sulfide minerals may not be well

Statistical Analysis. For all LCF analysis, the error bars on the contribution from a given standard are produced by Larch (25) in the model of IFEFFIT (55), and are the diagonal elements of the covariance matrix, i.e., the variances of the individual components, when the reduced χ^2 has been corrected to be equal to 1. This correction is necessary because the estimate of measurement error used in the χ^2 calculation does not include sample inhomogeneity and detector nonlinearity, which are the dominant sources of measurement error in XAS collected from modern synchrotron light sources. When calculating Fe(II)/Fe_{total} for each sediment sample, the LCF errors on the contribution from each standard are propagated, assuming a constant Fe(II) fraction for each standard.

Data Availability. All raw XAS data files and new PS75/56-1 data (²³²Th flux and Fe flux) are deposited in the Columbia University Academic Commons, with the persistent URL: <https://doi.org/10.7916/D8X07QHG>.

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